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Furan-2,3-diones 1a-c react with various hydrazines 2a-c under different conditions to yield the pyrazole-3-carboxylic acid-hydrazide 3a-d. Cyclocondensation reactions of $\mathbf{1 a}$ or $\mathbf{7}$ with phenylhydrazine lead to derivatives of pyrazolo[3,4- $d$ ] pyridazinones $\mathbf{6}$ and $\mathbf{8}$, respectively. The structures of all products were confirmed by elemental analysis, IR, ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectroscopic measurements.
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Introduction.
Pyrazole nucleus and its chemistry [1] has been the focus of high attention for more than three decades due to versatile biological activities of pyrazole derivatives appearing as anti-microbial [2], anti-viral [3], anti-tumor [4], anti-inflammatory [5], anti-histaminic [6], pesticidal [7], anti-fungal [8], against rheumatoid arthritis [9], anticonvulsant [10], anti-depressant [11], anti-pyretic [12] and commercially important dyestuffs [13] agents. Recently, reactions of cyclic oxalyl compounds have been reported to give substituted heterocyclic compounds [14]. The reactions of 4-benzoyl-5-phenylfuran-2,3-dione with several semicarbazones, ureas and their thio-anologues and oximes have been reported in different solvents and at various temperatures [15]. The general reactivity of 4-ben-zoyl-5-phenylfuran-2,3-dione and the mechanism of the reactions with NH-nucleophiles have recently been reviewed with semi-empirical (AM1 and PM3) calcula-
tions [16]. The reactions are generally initiated by nucleophilic attack of the nitrogen atom of semicarbazone, urea and hydrazines, on the furan ring $[17,18]$. The reaction of furan-2,3-dione 1a, obtained easily from dibenzoylmethane and oxalyl dichloride [14a], with various phenyl hydrazones and phenylhydrazine leads to pyrazole carboxylic acids and pyridazinones [19].

In the present study, we carried out the reaction of the furans-2,3-diones 1a,c with hydrazine derivatives 2a-c yielding pyrazole-3-carboxylic acid-hydrazide derivatives 3a-d, (Scheme 1). The structures of synthesized compounds were assigned on the basis of analytical results as well as spectroscopic data. Product 3a was obtained in $45 \%$ yield by treating 1a with 2,4-dinitrophenylhydrazine 2a and refluxing the mixture in toluene for 6 hours. The moderate yield of the reaction can be explained by the chemical behavior of furandiones 1a-c towards H -active nucleophiles. In compounds 1a-c carbon atoms C-2, C-3


1

| 1 | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | 3 | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| a | $p-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{4}-$ | p- $\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{4}$ - |  |  |  | $\mathrm{NO}_{2}$ |
| b | EtO- | Ph- | a | $p-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{4}-$ | $p-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{4}-$ |  |
| c | Ph- | Ph- | b | EtO- | Ph- | -Ph |
|  |  |  | c | Ph- | Ph- |  |
| 2 | $\mathrm{R}^{3}$ |  |  |  |  |  |
| a |  |  | d | Ph- | Ph- |  |
| b | -Ph |  |  |  |  |  |
| c |  |  |  |  |  |  |

Scheme 1
Schene 1


3


Ph-


and C-5 represent electrophilic sites of different reactivity and could be used for the construction reaction with nucleophiles [16a,17,20]. It should start with a nucleophilic attack of the nitrogen atoms lone pair electrons of $\mathbf{2 a - c}$ to the antibonding $\left(\pi^{*}\right)$ orbital at the C5 position of the furandione ring similar to a Michael-type addition. Simultaneous attack of H -active nucleophiles to both C-2 and $\mathrm{C}-3$ positions of the furan ring could convert furandiones 1a-c into starting materials; these materials are dibenzoylmethane and oxalic acid derivatives [15a]. The by-products formed this way are removed when the raw product is treated with diethyl ether. In the ir spectra of compound $\mathbf{3 a}$, the -NH absorption bands were found to be at about 3450-3300 $\mathrm{cm}^{-1}$, and the $\mathrm{C}=\mathrm{O}$ absorption was at $1700 \mathrm{~cm}^{-1}$. The ${ }^{1} \mathrm{H}-\mathrm{nmr}$ signals were found to be at 11.68 (b, 2H, -NH); 8.23-6.74 (m, 14, ArH), 3.87 and 3.63 ppm (q, $6 \mathrm{H}, 2 \mathrm{CH}_{3} \mathrm{O}$ ). The ${ }^{13} \mathrm{C}-\mathrm{nmr}$ signals were observed at $\delta$ 191.68 (t, PhCO), 173.67 (s, C=O), 148.90 (s, C3), 57.30 and $57.00 \mathrm{ppm}\left(\mathrm{q}, 2 \mathrm{CH}_{3} \mathrm{O}\right)$. Finally, the elemental analysis data along with spectroscopic data (see Experimental) confirm the structure of $\mathbf{3 a}$.
In a similar way, the reaction of the 4-benzoyl-1,5-diphenyl-1H-pyrazole-3-carboxylic acid chloride 4 [19b] with 2,4-dinitrophenylhydrazine $\mathbf{2 a}$ leads to form 4-ben-zoyl-1,5-diphenyl-1H-pyrazole-3-carboxylic acid 2,4-dinitrophenyl-hydrazide 5, (Scheme-2).

Scheme 3


8

## EXPERIMENTAL

Solvents were dried by refluxing with the appropriate drying agent and distilled before use. Melting points were determined on an Electrothermal 9200 apparatus and are uncorrected.

Scheme 2


Reaction of suitable vicinal dicarbonyl pyrazole derivatives with hydrazines in general are a convenient method to build the pyrazolo[3,4-d]pyridazine systems [17,19b,21]. Similiarly, the reaction of 4-(4-methoxyben-zoyl)-5-(4-methoxyphenyl)-furan-2,3-dione 1 a and 4-ben-zoyl-1-(2,4-dinitrophenyl)-5-phenyl-1H-pyrazole-3-carboxylic acid 7 [18] with phenylhydrazine 2b leads to the formation of 2,6-diphenyl-3,4-di-(4-methoxyphenyl)-2.6-dihydropyrazolo[3,4- $d$ ]pyridazin-7-one 6 and 2-(2,4-dini-trophenyl)-3,4,6-triphenyl-2,6-dihydropyrazolo[3.4-d]-pyridazin-7-one $\mathbf{8}$ in approximately 40-60 \% yields, (Scheme-3). In the ir spectra of compound 6, the $\mathrm{C}=\mathrm{O}$ absorption band was observed at $1690 \mathrm{~cm}^{-1}$. The ${ }^{1} \mathrm{H}-\mathrm{nmr}$ signals were observed at $\delta 7.78-6.56(\mathrm{~m}, 18 \mathrm{H}, \mathrm{ArH})$ and 3.75 and $3.73 \mathrm{ppm}\left(\mathrm{q}, 6 \mathrm{H}, 2 \mathrm{CH}_{3} \mathrm{O}\right)$ and the ${ }^{13} \mathrm{C}-\mathrm{nmr}$ signals at $\delta 162.18$ (s, C=O); 149.52 (s, C-7a), 145.36 ( $\mathrm{s}, \mathrm{C} 4$ ), $142.03(\mathrm{~s}, \mathrm{C} 3), 57.29$ and $56.75 \mathrm{ppm}\left(\mathrm{q}, 2 \mathrm{CH}_{3} \mathrm{O}\right)$.

Microanalyses were performed on a Carlo Erba elemental analyser, Model 1108; the results agree favorably with the calculated values. The IR spectra were recorded on a Shimadzu Model 435 V-04 spectrometer, using potassium bromide discs. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Gemini-Varian 200 instrument. The chemical shifts are reported in ppm from tetramethylsilane and are given in $\delta$ (ppm).
4-(4-Methoxybenzoyl)-5-(4-metoxyphenyl)-1-(2,4-dinitro-phenyl)-1H-pyrazole-3-carboxylic acid-2,4-dinitro phenylhydrazide (3a).

4-(4-Methoxybenzoyl)-5-(4-methoxyphenyl)furan-2,3-dione 1a $(0.2 \mathrm{~g}, 0.59 \mathrm{mmol})$ and 2,4-dinitrophenylhydrazine $2 \mathbf{2 a}$ ( 0.234 $\mathrm{g}, 1.18 \mathrm{mmol})$ were refluxed in benzene for 4.5 hours. The solvent was evaporated and the remaining oily residue was treated with ether to give the crude product which was recrystallized from acetic acid and allowed to dry on $\mathrm{P}_{2} \mathrm{O}_{5} ; 45 \%$ yield $(0.19 \mathrm{~g})$; m.p. $229^{\circ} \mathrm{C}$; ir ; 3450-3300 (broad, N-H), $1700 \mathrm{~cm}^{-1}$ (C=O); ${ }^{1} \mathrm{H}-$ $\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 11.68(\mathrm{~b}, 2 \mathrm{H},-\mathrm{NH}), 8.23-6.74(\mathrm{~m}, 14 \mathrm{H}, \mathrm{ArH})$, 3.87 and $3.63\left(\mathrm{q}, 6 \mathrm{H}, 2 \mathrm{CH}_{3} \mathrm{O}\right) ;{ }^{13} \mathrm{C}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 191.68(\mathrm{t}$,

PhCO ), 173.67 ( $\mathrm{s}, \mathrm{C}=\mathrm{O}$ ), 148.90 ( $\mathrm{s}, \mathrm{C}-3$ ), 57.30 and 57.00 (q, $2 \mathrm{CH}_{3} \mathrm{O}$ ).

Anal. Calcd. for $\mathrm{C}_{31} \mathrm{H}_{22} \mathrm{~N}_{8} \mathrm{O}_{12}$ : C, $53.29 ; \mathrm{H}, 3.15 ; \mathrm{N}, 16.05$. Found; C, 53.36; H, 3.07; N, 16.10.
4-(Ethoxycarbonyl)-1,5-diphenyl-1 H -pyrazole-3-carboxylic Acid Phenylhydrazide (3b).

Ethyl-4,5-dioxo-2-phenyl-4,5-dihydrofuran-3-carboxylate 1b $(0.5 \mathrm{~g}, 2.03 \mathrm{mmol})$ and phenyl hydrazine $\mathbf{2 b}(0.44 \mathrm{~g}, 0.4 \mathrm{ml}, 4.07$ mmol ) were refluxed in benzene for 1 hour or stirred in benzene at room temperature for 24 hours. The solvent was evaporated and the remaining oily residue was treated with $n$-hexane and ether and stirred for 24 hours to give the colorless product which was recrystallized from $n$-butyl alcohol and allowed to dry on $\mathrm{P}_{2} \mathrm{O}_{5} ; 67 \%$ yield ( 0.54 g); m.p. $204{ }^{\circ} \mathrm{C}$; ir; 3450-3300 (b, N-H), 1680 (amide $\mathrm{C}=\mathrm{O}$ ), 1661 (ester $\mathrm{C}=\mathrm{O}$ ), $1620-1480 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}-\mathrm{nmr}$ $\left(\mathrm{CDCl}_{3}\right): \delta 10.62$ and $10.60(2 \mathrm{H},-\mathrm{NH}) ; 8.02-6.57(\mathrm{~m}, 14 \mathrm{H}, \mathrm{ArH})$; $4.26\left(\mathrm{O}-\mathrm{CH}_{2}\right), 1.20\left(\mathrm{CH}_{3}\right) ;{ }^{13} \mathrm{C}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 163.61(\mathrm{~N}-\mathrm{C}=\mathrm{O})$; 161.30 ( $\mathrm{O}-\mathrm{C}=\mathrm{O}$ ); 150.21 (s, C-5); 143.20 (C-3).

Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{18} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 69.29; H, 4.75; N, 14.10. Found; C, 69.19; H, 4.96; N, 13.97.
4-Benzoyl-5-phenyl-1-(4-nitrophenyl)-1 H -pyrazole-3-carboxylic Acid- 4-Nitrophenylhydrazide (3c).

4-Benzoyl-5-phenylfuran-2,3-dione $1 \mathbf{c}(0.5 \mathrm{~g}, 1.798 \mathrm{mmol})$ was heated in toluene and allowed to cool to room temperature at which time 4-nitrophenyl hydrazine $2 \mathrm{c}(0.55 \mathrm{~g}, 3.59 \mathrm{mmol})$ was added and the mixture was stirred at room temperature for 2 days. The solvent was evaporated and the remaining oily residue was treated with diethyl ether. The yellow precipitate was collected by filtration and washed in a mixture of petroleum ether and cyclohexane and allowed to dry on $\mathrm{P}_{2} \mathrm{O}_{5} ; 45 \%$ yield ( 0.35 g ); m.p. $197{ }^{\circ} \mathrm{C}$; ir; 3450-3250 (b, N-H), $1720 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 11.05$ and $10.80(2 \mathrm{H},-\mathrm{NH}) ; 8.51-$ 6.49 (m, 13H, Ar-H); ${ }^{13} \mathrm{C}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 193.16(\mathrm{t}, \mathrm{PhCO})$, 166.74 (s, C-7), 151.16 (s, C-5), 150.90 (s, C-3); 140.33-112.22 ( m , Aromatic C).
Anal. Calcd. for $\mathrm{C}_{25} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{5}$ : C, $63.50 ; \mathrm{H}, 3.65 ; \mathrm{N}, 15.33$. Found; C, 63.65; H, 3.83; N, 15.01.

4-Benzoyl-5-phenyl-1-[2,4-dinitrophenyl]-1 H -pyrazole-3-carboxylic Acid 2,4-Dinitro phenylhydrazide (3d).
4-Benzoyl-5-phenylfuran-2,3-dione $1 \mathrm{c}(0.5 \mathrm{~g}, 1.798 \mathrm{mmol})$ and 2,4-dinitrophenylhydrazine $2 \mathbf{2 a}(0.712 \mathrm{~g}, 3.60 \mathrm{mmol})$ were refluxed in toluene for 6 hours. The solvent was evaporated and the remaining oily residue was treated with diethyl ether. The yellow crude by-product that did not dissolved in diethyl ether was removed by filtration. The precipitate was collected by filtration and washed in hot ethyl alcohol and recrystallized from acetic acid and allowed to dry on $\mathrm{P}_{2} \mathrm{O}_{5} ; 45 \%$ yield ( 0.52 g ); m.p. 158$159{ }^{\circ} \mathrm{C}$; ir: $3500-3150(\mathrm{~b}, \mathrm{~N}-\mathrm{H}), 1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}-\mathrm{nmr}$ $\left(\mathrm{CDCl}_{3}\right): \delta 9.40$ and $9.28(2 \mathrm{H},-\mathrm{NH}), 7.32-6.97(\mathrm{~m}, 16 \mathrm{H}, \mathrm{ArH})$; ${ }^{13} \mathrm{C}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 196.54(\mathrm{t}, \mathrm{PhCO}), 159.84(\mathrm{~s}, \mathrm{C}-7), 157.29(\mathrm{~s}$, C-5), 149.97-116.00 (m, Aromatic C).

Anal. Calcd. for $\mathrm{C}_{29} \mathrm{H}_{18} \mathrm{~N}_{8} \mathrm{O}_{10}$ : C, 54.55; H, 2.82; N, 17.55. Found; C, 54.27; H, 2.99; N, 17.56.
4-Benzoyl-1,5-diphenyl-1 H -pyrazole-3-carboxylic Acid 2,4Dinitrophenylhydrazide (5).

4-Benzoyl-1,5-diphenyl-1 H -3-pyrazole carboxylic acid chloride $4(0.5 \mathrm{~g}, 1.29 \mathrm{mmol})$ and 2,4-dinitrophenylhydrazine 2a
$(0.256 \mathrm{~g}, 1.29 \mathrm{mmol})$ were refluxed in xylene for 10 hours. The solvent was evaporated and the remaining oily residue was treated with diethyl ether to give an orange crude product which was recrystallized from ethyl alcohol and allowed to dry on $\mathrm{P}_{2} \mathrm{O}_{5} ; 30 \%$ yield ( 0.21 g ); m.p. $167^{\circ} \mathrm{C}$; ir: 3500-3150 (b, N-H), $1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 9.45$ and $9.19(2 \mathrm{H},-\mathrm{NH})$, 7.32-6.97 (m, 18H, ArH); ${ }^{13} \mathrm{C}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 193.44(\mathrm{t}, \mathrm{PhCO})$, 162.09 (s, C-7), 150.63 (s, C-5), 149.97-116.00 (m, Aromatic C).

Anal. Calcd. for $\mathrm{C}_{29} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{O}_{6}$ : C, 63.50; H, 3.65; N, 15.33. Found; C, 63.63; H, 3.82; N, 15.06.
2,6-Diphenyl-3,4-di-(4-methoxyphenyl)-2.6-dihydropyra-zolo[3,4- $d$ ]pyridazin-7-one (6).

4-(4-Methoxybenzoyl)-5-(4-methoxyphenyl)furan-2,3-dione $\mathbf{1 a}(0.3 \mathrm{~g}, 0.89 \mathrm{mmol})$ and phenylhydrazine $\mathbf{2 b}(0.192 \mathrm{~g}, 0.174$ $\mathrm{ml}, 1.78 \mathrm{mmol})$ were refluxed in benzene for 5 hours. The solvent was evaporated and the remaining oily residue was treated with diethyl ether to give a crude product that was recrystallized from methanol and allowed to dry on $\mathrm{P}_{2} \mathrm{O}_{5}$. Compound $\mathbf{6}$ was obtained in $40 \%$ yield $(0.176 \mathrm{~g})$; m.p. $234{ }^{\circ} \mathrm{C}$; ir: $1690 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ;{ }^{1} \mathrm{H}-$ $\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 7.78-6.56(\mathrm{~m}, 18 \mathrm{H}, \mathrm{ArH}), 3.75-3.73(\mathrm{q}, 6 \mathrm{H}$, $2 \mathrm{CH}_{3} \mathrm{O}$ ); ${ }^{13} \mathrm{C}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta 162.18(\mathrm{~s}, \mathrm{C}=\mathrm{O}), 149.52(\mathrm{~s}, \mathrm{C}-7 \mathrm{a})$, 145.36 (s, C-4), 142.03 (s, C-3), 129.04 (s, C-3a), 57.29 and $56.75\left(\mathrm{q}, 2 \mathrm{CH}_{3} \mathrm{O}\right)$.

Anal. Calcd. for $\mathrm{C}_{31} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{3}$ : C, 74.40; H, 4.80; N, 11.20. Found; C, 74.15; H, 4.85; N, 11.23.
2-(2,4-Dinitrophenyl)-3,4,6-triphenyl-2,6-dihydropyrazolo-[3,4- $d$ ]pyridazin-7-one (8).

4-Benzoyl-1-(2,4-dinitrophenyl)-5-phenyl-1 H -3-pyrazole carboxylic acid $7(0.5 \mathrm{~g}, 1.09 \mathrm{mmol})$ and phenylhydrazine $\mathbf{2 b}(0.12 \mathrm{~g}$, $0.11 \mathrm{ml}, 1.11 \mathrm{mmol}$ ) were refluxed in xylene for 9 hours. The solvent was evaporated and the remaining oily residue was treated with diethyl ether to give a yellow crude product which was recrystallized from $n$-butyl alcohol and allowed to dry on $\mathrm{P}_{2} \mathrm{O}_{5}$, resulting in $60 \%$ yield ( 0.35 g ); m.p. $279^{\circ} \mathrm{C}$; ir: $1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$; ${ }^{1} \mathrm{H}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): ~ \delta 8.77-6.97(\mathrm{~m}, 18 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}-\mathrm{nmr}\left(\mathrm{CDCl}_{3}\right): \delta$ 159.84 (s, C=O); 149.25 (s, C-7a), 145.48 (s, C-4), 142.88 (C$\mathrm{NO}_{2}$ ), 141.59 (s, C-3), 136.36 ( $\mathrm{N}-\mathrm{Ph}$ ), 119.50 (s, C-3a).

Anal. Calcd. for $\mathrm{C}_{29} \mathrm{H}_{18} \mathrm{~N}_{6} \mathrm{O}_{5}$ : C, 65.66 ; H, 3.42; $\mathrm{N}, 15.84$. Found; C, 65.88; H, 3.52; N, 15.89.

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